

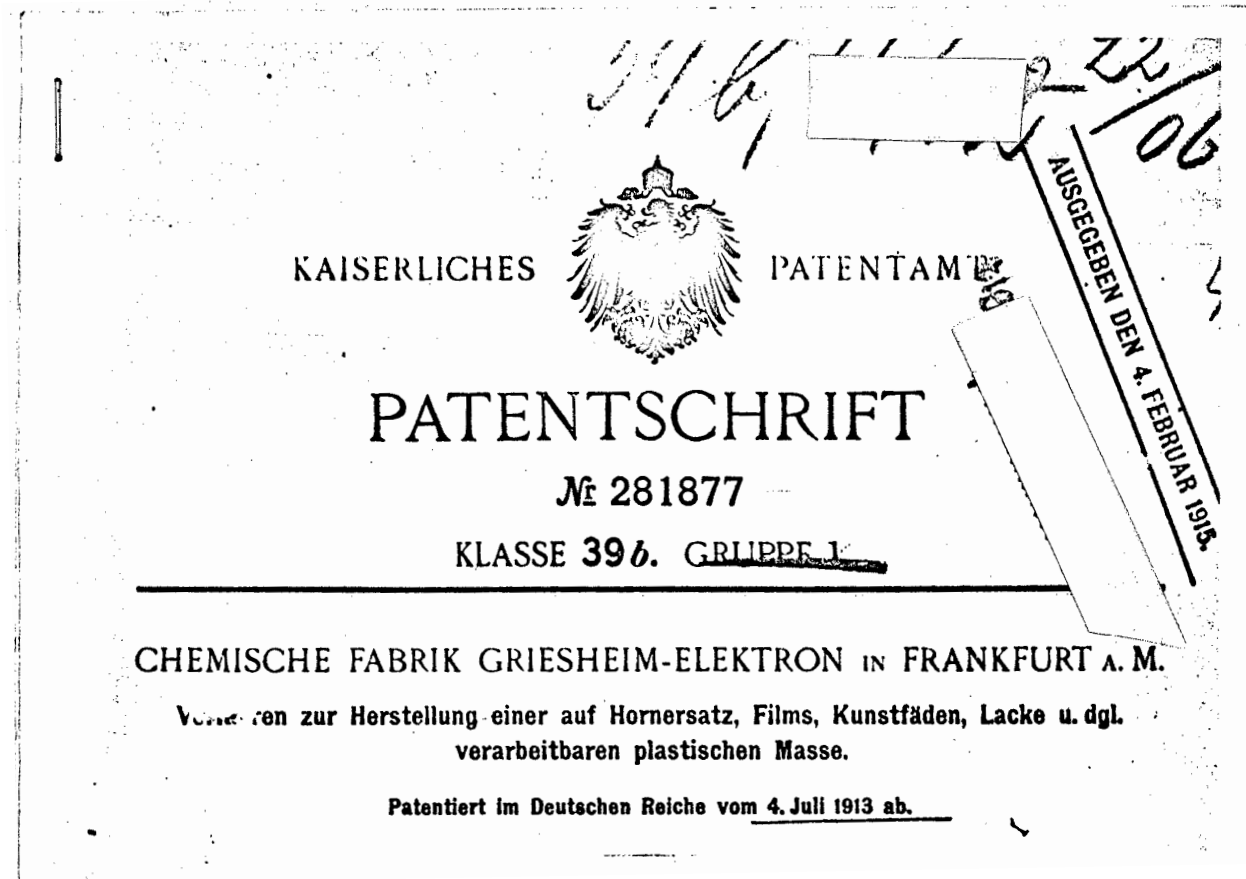


Metal soaps under the viewpoint of sustainability - One Vision

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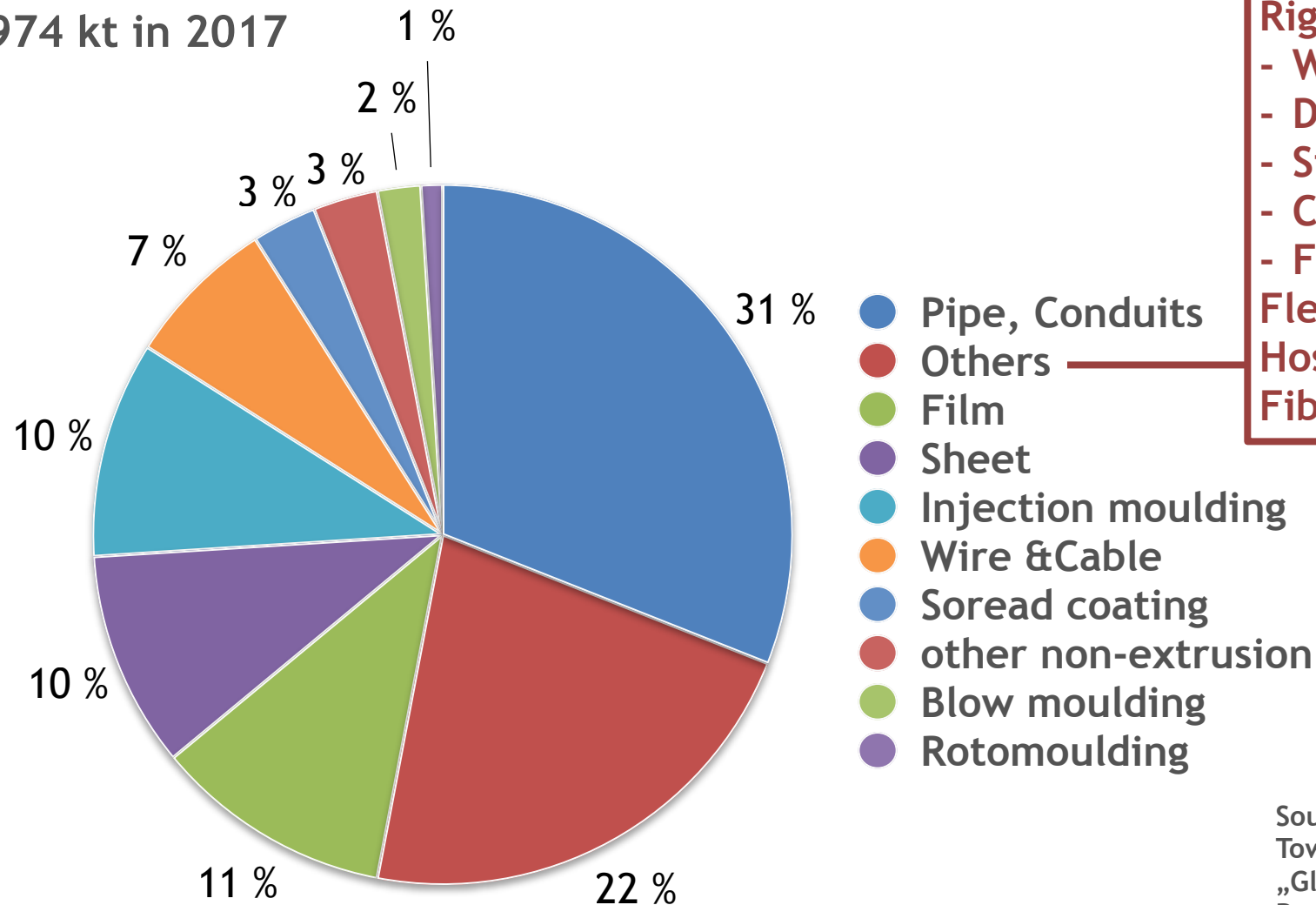
About PVC

1st patent claiming the industrial production of PVC filed by Klatte, Zacharias and Rollett on 04.07.1913



Global PVC Consumption by Main Applications

44'974 kt in 2017

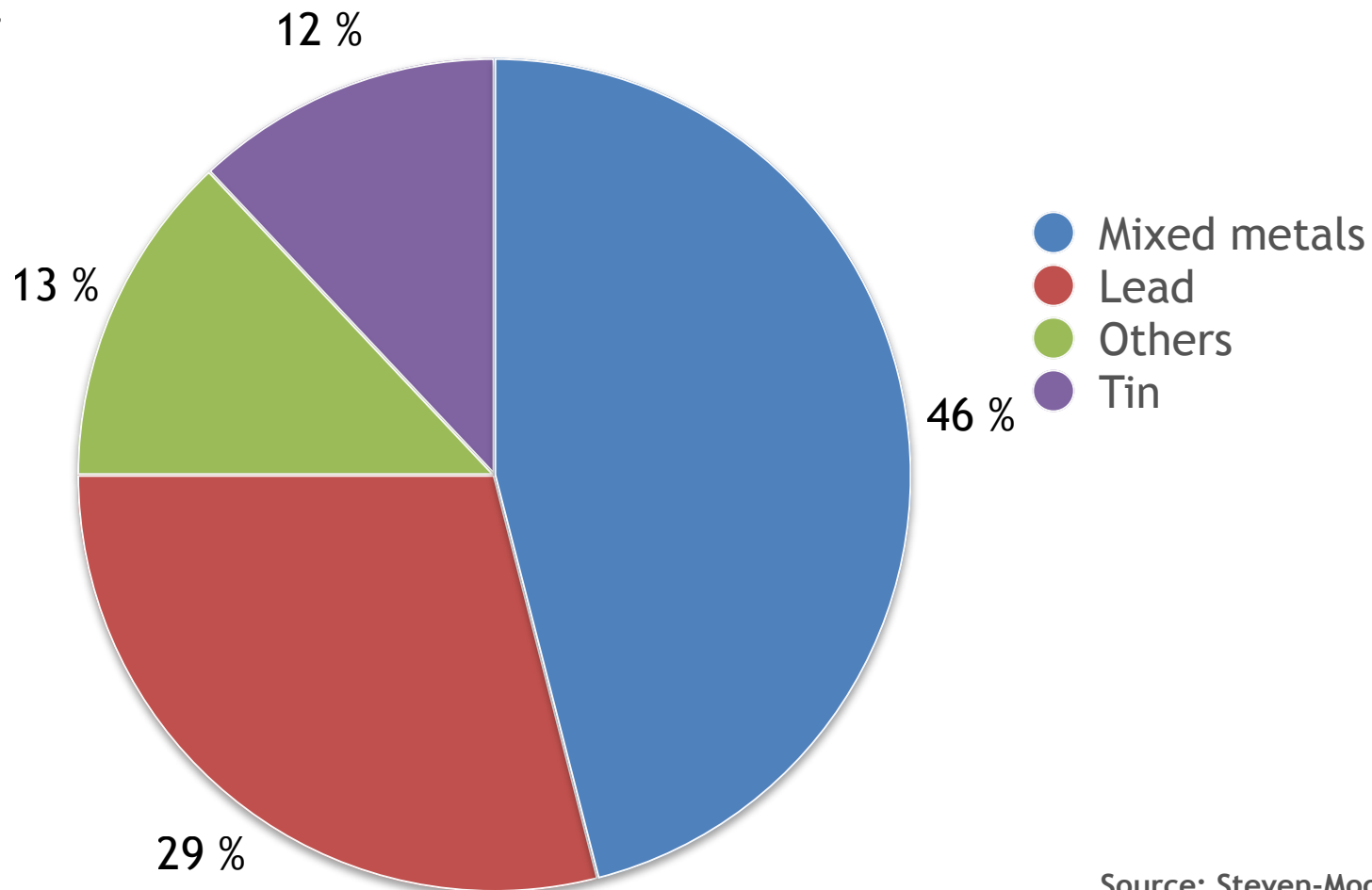


Rigid profiles
 - Window
 - Door
 - Siding
 - Ceiling
 - Flooring
Flexible profile
 Hose & tube
 Fiber

Source: Steven-Moore,
 Townsend Solutions
 „Global Trends in PVC
 Resin Applications and
 Additives
 Usage“ 15.03.2018

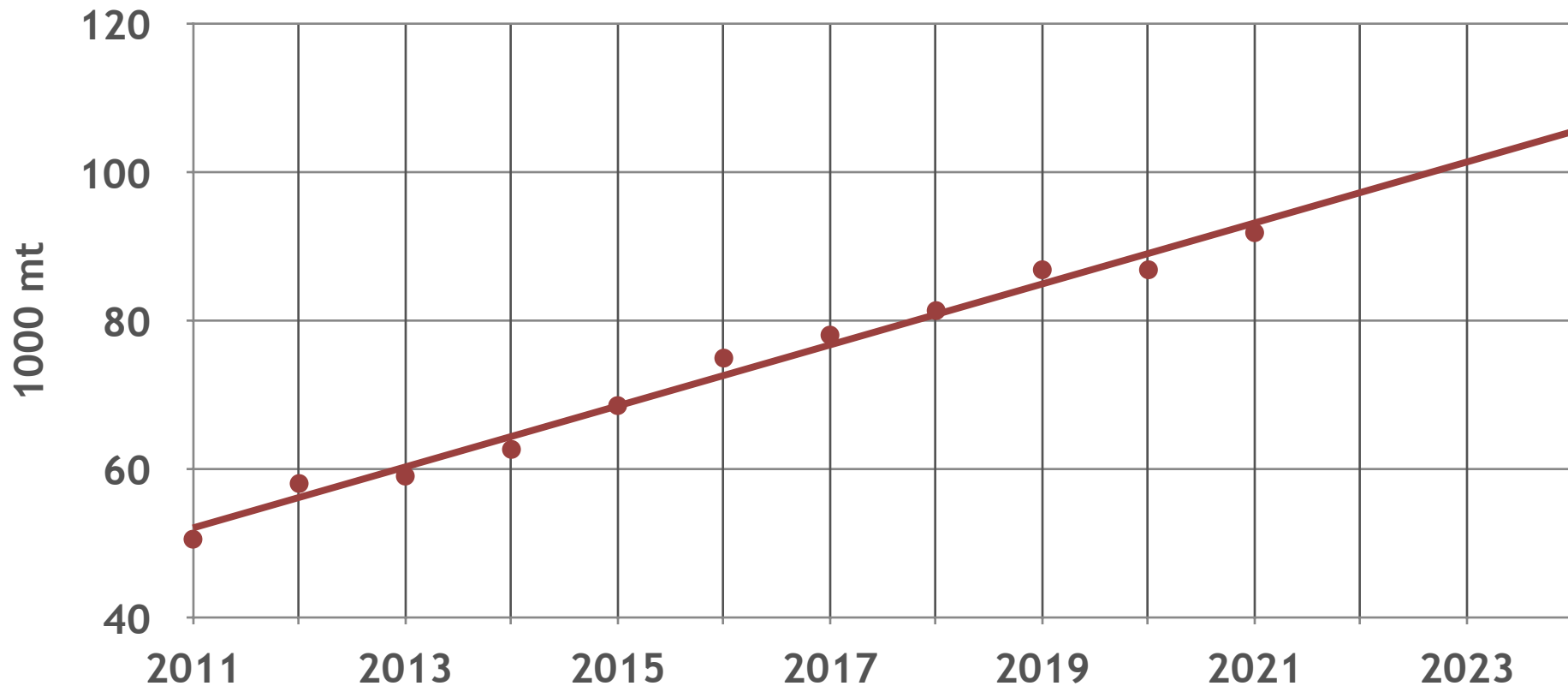
Global Heat Stabilizer Consumption by Type

1'126 kt in 2017



Source: Steven-Moore,
Townsend Solutions
„Global Trends in PVC
Resin Applications and
Additives
Usage“ 15.03.2018

Annual PVC stabiliser consumption in India



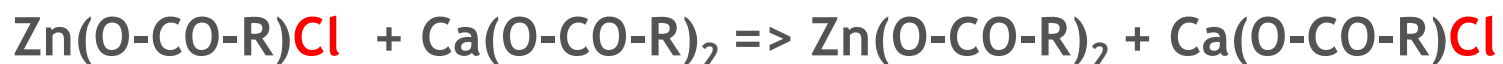
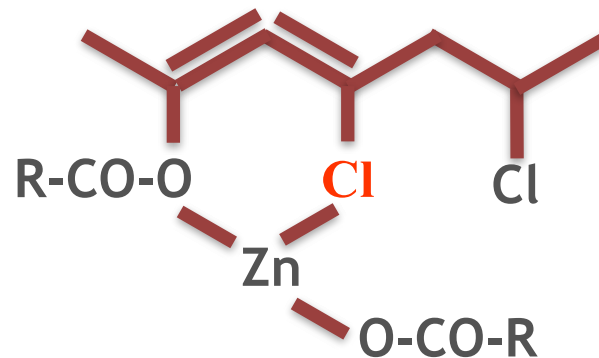
About PVC stabilisers

- Substitution of unstable chlorine atoms in PVC chain
- Neutralisation of HCl
- Shortening the pollen sequences
- Avoiding autoxidation
- Guaranteeing optimal processing and best performance of final product during its lifetime
- Systems based on:
 - Lead
 - Tin
 - Antimony
 - Barium
 - Calcium
 - Cadmium
 - Zinc
 - Lanthanum

How does a metal soap „work“ in PVC?

■ Metals: Mg, Ca, Ba, Zn, Cd, Pb, (K)

■ Acids: Stearic, Oleic, Benzoic, Lauric, 12-Hydroxystearic, Octanoic, 2-Ethylhexanoic, Montanoic



Production of metal soaps

■ Principles:



Me : metal

HAc: acid

Baseline line analysis of metal soap production

- Direct reaction of metal (hydr)oxide and acid in water as reaction media without any by-product beside water
- Direct and dry conversion of acid and metal (hydr)oxide without solvent in a high speed mixer without any by-product beside water
- Melt process at temperatures depending on the melt temperature of the metal soap and the acid without any by-product beside water
- Two step process in water:
 - Formation of alkaline (mainly Sodium) salt
 - Addition of a solution of the soluble metal salt (e.g. Calcium chloride)
 - By-product: alkaline salt (e.g. Sodium chloride) dissolved in water

Introduction in TNS Framework - Scientific background

- The earth is a closed system regarding matter. (Nothing disappears. Nothing is added.)
- All geological processes are slow. (Therefore everything that is on the surface of the earth stays permanently there.)
- The earth is an open system regarding energy.
- 1st law of thermodynamics; the conservation of energy. (Energy can neither be created nor destroyed, only converted into other types of energy. Matter can be converted into energy, heat and movement.)
- 2nd law of thermodynamics, the introduction of entropy. (A system is all the more stable the greater the disorder in this system.)
- Photosynthesis is the "motor of all life" on earth.
- From the local to the global.
From a few, large sources to diffuse, scattered sources.
From short maturities to long and delayed.

Introduction - Four System Conditions (SC) of TNS

In a sustainable society nature is not subject to systematically increasing...

1
... concentrations
of substances
extracted from
the earth's
crust

2
... concentrations
of substances
produced by
society

3
... degradation
by physical
means

4
... and, in that
society, people
are not subject
to conditions
that systemically
undermine their
capacity to meet
their needs

System Conditions 1 of TNS

In a sustainable society nature is not subject to systematically increasing...

...
1
 concentrations
 of substances
 extracted from
 the earth's
 crust

Element	mg/kg in the earth crust	kt/year Erosion (ϵ)	kt/year Mining (μ)	kt/year Fossil fuels (f)	($\mu+f/\epsilon$)
Ca	50·10 ⁶	750·10 ⁶			<1.000
Na	23·10 ⁶	345·10 ⁶			<1.000
Si	310000.0	4700000.0	4600.0	95000.0	0.021
Mg	9000.0	140000.0	3100.0	690.0	0.028
Al	72000.0	1100000.0	18000.0	34000.0	0.048
Ti	2900.0	44000.0	2500.0	1700.0	0.096
K	15000.0	230000.0	24000.0	340.0	0.110
La	32.0	480.0	12.5	0.3-258.2	0.027-0.564
Ba	400.0	6000.0	4532.0	117.8	>0.800
Fe	26000.0	390000.0	540000.0	34000.0	1.500
P	430.0	6500.0	21000.0	1700.0	3.500
S	1600.0	33000.0	58000.0	100000.0	3.700
Cd	0.4	5.3	20.0	3.4	3.900
C	25000.0	780000.0	-	5400000.0	6.400
Zn	60.0	910.0	7300.0	260.0	8.300
Sn	1.3	20.0	210.0	5.7	11.000
Pb	19.0	290.0	3300.0	85.0	12.000
Al	72000.0	1100000.0	18000.0	34000.0	0.048
Ti	2900.0	44000.0	2500.0	1700.0	0.096

Compelling vision of fully sustainable metal soaps

SC 1

All used substances in a metal soap must have the ratio $(\mu+f/\epsilon) < 1$.

Energy source must be renewable!

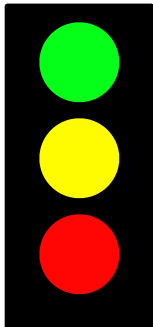
Raw material must be either from renewable sources or based on recycling.

Metal soap itself and its decomposition products must be non-hazardous and non-toxic.

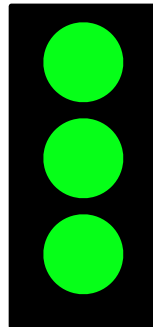
SC 3

The mining of raw materials must be in that way that the natural conditions are not disturbed and later can be restored.

Storage of waste products must be in that way that the natural conditions are not disturbed and later can be restored.



Today



Vision

SC 2

Organic raw materials used during production of metal soap must be based on renewable resources and degradable.

Organic parts of metal soap, which can migrate, must be degradable.

Emissions and waste of production and during the use of metal soap must be either degradable or recyclable. Recycling process must be clean.

There must not be any contribution to the progressive buildup of chemicals and compounds produced like dioxins...

SC 4

A reliable performance is required to satisfy the needs of life. The metal soaps may not have any negative influence during recycling of "end of life" products and on the performance of the new product, if:

- based on the "end of life" product a new analogous product is produced

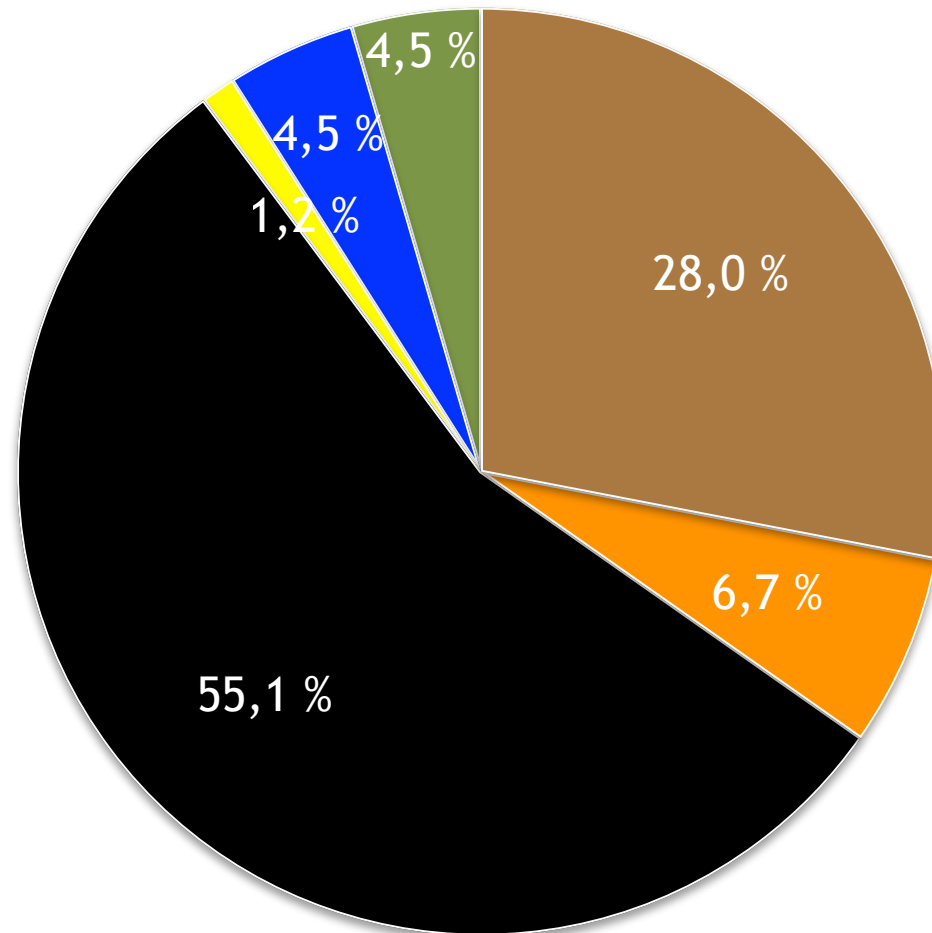
- recycled material is mixed with fresh material

Social value = social benefit - social costs → final products may not have any negative effect on wealth of people or their environment

The production of metal soaps must guarantee the economical survival of the producer and user.










Baseline line analysis of metal soap production - use of primary energy in India in 2020

● Oil ● Natural gas ● Coal ● Nuclear ● Hydro ● Renewables



Source: <https://www.bp.com/content/dam/bp/business-sites/en/global/corporate/pdfs/energy-economics/statistical-review/bp-stats-review-2021-india-insights.pdf>; accessed 31.10.2022













Baseline line analysis of metal soap production - metal (hydr)oxides

	SC 1	SC 2	SC 3	SC 4
MgO				
Ca(OH) ₂				
Ba(OH) ₂				
ZnO				
PbO				
CdO				















Baseline line analysis of metal soap production - acids

Acid	SC 1	SC 2	SC 3	SC 4
Stearic	●	●	●	●
Oleic	●	●	●	●
Lauric	●	●	●	●
12-HOST	●	●	●	●
Caprylic	●	●	●	●
Benzoic	●	●	●	●
2-Ethyl hexanoic	●	●	●	●

Baseline line analysis of metal stearates production - other parameter

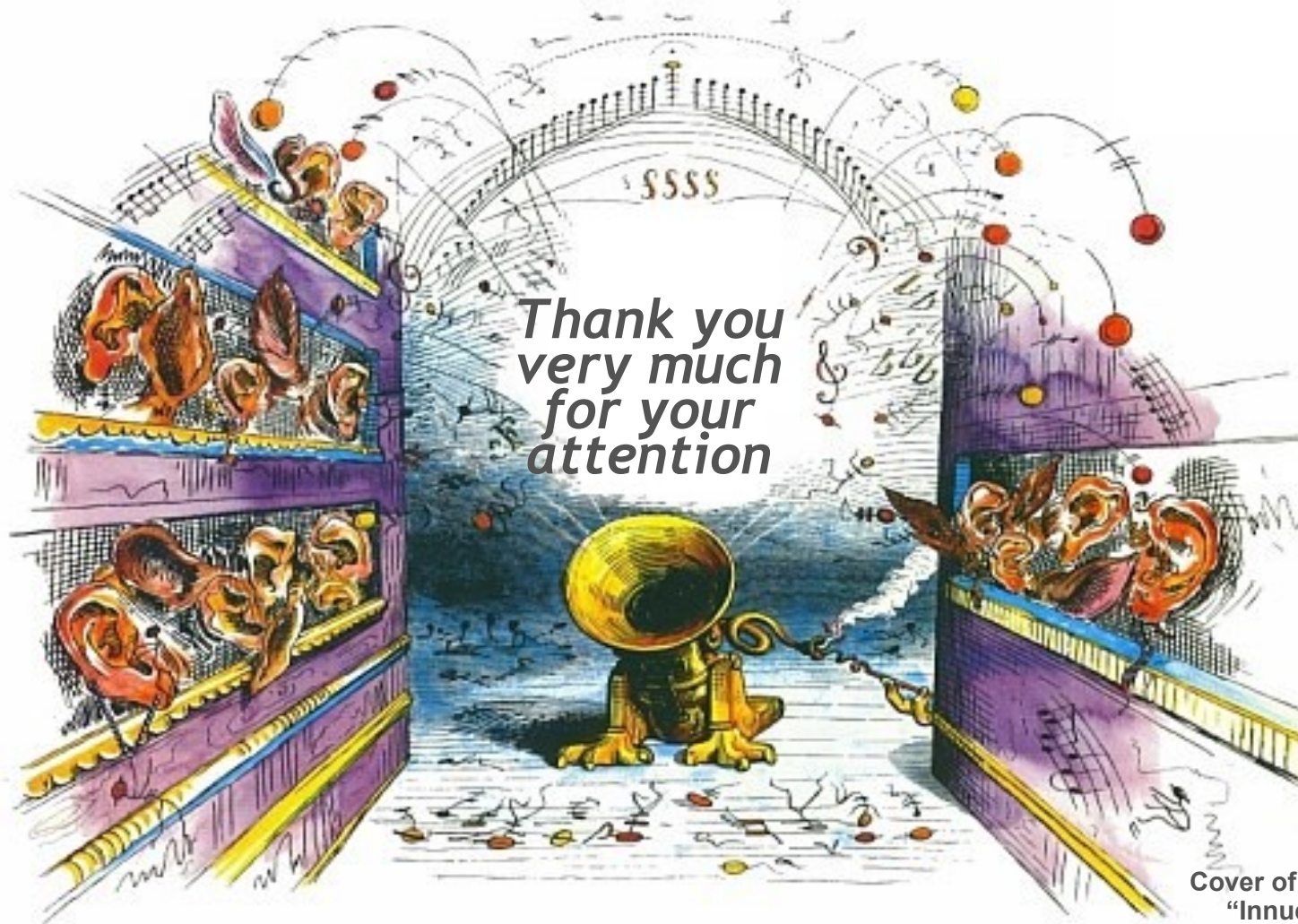
	SC 1	SC 2	SC 3	SC 4
Water				
Energy				
Packaging	?	?	?	?
Production plant				
Waste	?	?	?	?
Transportation	?	?	?	?

Baseline line analysis of metal stearates production - end of service time

	SC 1	SC 2	SC 3	SC 4
In land fill	 			
In combustion	 			
In recycling				

Summary

- Metal soaps (Ca, Mg, Al) have already a high potential to become fully sustainable!
- Exceptions are all soaps based on toxic metals like Lead and Cadmium
- Another exception is Zinc which will become sustainable if the PVC product is recycled.
- Metals soaps based on renewable acids have the highest sustainability potential today.
- Metal soaps based on fossil fuel have the lowest sustainability potential today.



*Thank you
very much
for your
attention*

Cover of QUEEN's Vinyl
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